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Element Symbol Atomic . Number Mass. Number Number of. neutrons nitrogen-15 8 Beryllium-9 4 3. Use the following information to determine the atomic mass of chlorine. Two isotopes are known: chlorine-35 (mass = 34.97 amu) and chlorine-37 (mass = 36.97 amu). The relative abundance's are 75.4% and 24. 6%, respectively. 4.

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Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these three

Chapter 4 Atomic Structure - Henry County School District

Section 4.3 CHAPTER Name 4 51 protons, 51 electrons, 72 neutrons 4He 2 22Ne 10 21. uranium-234 20. cesium-133 234U 92 133Cs 55 mass number atomic number 24Mg 12 d. all of the above. b. smaller than the mass of a neutron. 22 Chemistry: Matter and Change • Chapter 4 Study Guide for Content Mastery d. both the relative abundance and the mass of ...

Ch 4 study guide answers

This Chapter 4 Worksheet-Part B Atomic Structure Worksheet is suitable for 10th - 12th Grade. In this atomic structure worksheet, students answer ten questions about the structure of atoms including their subatomic particles, what atomic number and mass number indicate, what

Chapter 4 Atomic Structure Worksheet Answers

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