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Terms in this set (25) Chemical formula.
indicates the relative number of atoms
of each kind in a chemical compound.
Reveals the number of atoms of each
element contained in a single molecule

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Aluminum sulfite Formula $PbCl_2$
(NH_4HPO_4 Hydroiodic 54 acid SECTION
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203-206) VOCABULARY Write true or
false for each statement. 1.

Modern Chemistry • CHAPTER 7
CHAPTER 7 REVIEW Chemical Formulas
and Chemical Compounds SECTION 1

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Answer Key

SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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Answer Key **Review Answer Key**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions

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can be attached to the iron atom. (c) the charge on each Fe ion.

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Chromium is $+4$ and each oxygen is -2 . 56 SECTION 7-2 REVIEW MODERN CHEMISTRY HRW material copyrighted under notice appearing earlier in this work. Name Date Class CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 7-3 SHORT ANSWER Answer the following questions in the space provided. 1.

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Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW Chemical Formulas
and Chemical Compounds SECTION 1
SHORT ANSWER Answer the following
questions in the space provided. 1. c In a
Stock system name such as iron (III)
sulfate, the Roman numeral tells us (a)

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how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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CHAPTER 7 REVIEW Chemical Formulas
and Chemical Compounds MIXED
REVIEW SHORT ANSWER Answer the

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following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

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Compounds

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions

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can be attached to the iron atom.

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Chemical Formulas and Chemical
Compounds. CHAPTER 7 REVIEW
Chemical Formulas and Chemical

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Compounds SECTION 4 SHORT ANSWER
Answer the following questions in the
space provided. 1. Write empirical
formulas to match the following
molecular formulas: CH₃O 2 a.

Modern Chemistry Chapter 7 Review Answers Section 4

The Arrangement of Electrons in Atoms

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chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with how electrons are arranged in atoms.

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