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The Composition of
Ocean Water 4-3
Enrich On average, one
kilogram of ocean
water contains about
35 grams of salts. That
is, salts make up about
3.5% of ocean water.

Though sodium

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chloride is the most abundant and familiar salt in seawater, a variety of other salts are also dissolved in seawater.

Ocean Water Chemistry 4-3 Review and Reinforce

> Massive emissions of carbon dioxide into the atmosphere have an impact on the chemical and biological processes in the ocean.

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The warming of ocean water could lead to a - destabilization of solid methane deposits on the sea floor. Because of the excess CO₂, the oceans are becoming more acidic.

Ocean chemistry « World Ocean Review

Seawater is a complex mixture of 96.5 percent water, 2.5 percent salts, and smaller amounts of other substances, including

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dissolved inorganic and organic materials, particulates, and a few atmospheric gases.

Clear ocean water near a beach on Grand Bahama Island in The Bahamas. © Philip Coblentz—Digital Vision/Getty Images

**seawater |
Composition,
Salinity,
Distribution, & Facts**

...

than cold water, warm

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water forms only a thin layer on the ocean surface. The deeper you descend into the ocean, the colder. and . more dense. the water becomes. Gases in Ocean Water. Just as land organisms use gases found in air, ocean organisms use gases found in ocean water. Two gases that ocean organisms use are . carbon dioxide. and . oxygen. Carbon dioxide is more

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plentiful in the oceans
than in the air.

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In this case, the hydrate only forms at great depths. If the water is very cold, the methane hydrates could conceivably form in shallower water depths, or even at atmospheric pressure. In the open ocean, where the average

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bottom-water temperatures are around 2 to 4 degrees Celsius, methane hydrates occur starting at depths of around 500 metres.

Climate change and methane hydrates « World Ocean Review

Chemical composition of seawater; Salinity and the major constituents OCN 623 – Chemical

Oceanography Salt

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dissolved in ocean water alters the properties of water
Freezing point of seawater is $\sim -1.8^{\circ}\text{C}$
Density continually increases to freezing point Drives the circulation mode of the oceans -completely different from freshwater in lakes

Chemical composition of seawater; Salinity and the major ...

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Bromine, iodine, and boron also are constituents of sea water, and salts containing these elements can be expected to accompany other salts in rainwater derived from the oceans. Data on bromine and iodine are scanty. Hutchinson (1957) suggests that these constituents will show a seasonal variation similar to that of chloride ions.

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**Rainwater as a
Chemical Agent of
Geologic Processes
A Review**

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Chemistry of Sea
water..... If suspended

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solid material of either organic or inorganic origin is excluded, sea water may be considered as an aqueous solution containing a variety of dissolved solids and gases. Determination of the chemical nature and concentrations of the dissolved substances is difficult for

Chemistry of Sea water - California

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Photochemical reactions involving DOC can influence the chemistry of vital trace nutrients such as iron, and, even at dissolved concentrations on the order of one nanomole/kg (1×10^{-9} mole/kg), dissolved organic substances in the upper ocean waters are capable of greatly altering the bioavailability of essential trace

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nutrients, as, for example, copper and zinc.

Seawater - Dissolved organic substances | Britannica

Water is present in the marine environment as a liquid, a solid, and a gas regulated by temperature. Heat causes the water molecules to move. The greater the heat, the faster they move until the movement

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causes the hydrogen bonds to break converting liquid water to gas. Water turns to vapor at 100° C.

Ocean Chemistry ~ MarineBio Conservation Society

Ocean chemistry, also known as marine chemistry, is influenced by plate tectonics and seafloor spreading, turbidity currents, sediments,

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pH levels, atmospheric constituents, metamorphic activity, and ecology. The field of chemical oceanography studies the chemistry of marine environments including the influences of different variables.

Ocean chemistry - Wikipedia

Water Chemistry 101
for your Reef Aquarium
- Duration: 6:16. ...

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Deep Sea Drinking
Water - drink to your
health - Duration: 6:20.
keith manning 13,033
views. 6:20. Language:
English

Ocean Water Chemistry.mp4

Ocean acidification
occurs when CO₂ is
absorbed into the
water at a high rate. It
reacts with water
molecules (H₂O) to
form carbonic acid
(H₂CO₃). This

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compound then breaks down into a hydrogen ion (H^+) and bicarbonate (HCO_3^-). The presence of all these hydrogen ions is what decreases the pH, or acidifies the ocean.

The Chemistry Of Ocean Acidification | Climate Interpreter

water liberates ions from earth materials and adds them to the ocean. Sea spray from waves removes ions

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from the ocean when it deposits a film of salts on the land. Other methods of ion addition and removal also are depicted.

Ocean Chemical Processes - river, sea, oceans ... - Water

Subject: Ocean Physics, Ocean Biology, Ocean Chemistry,
Inquiry: Grade: 6 - 12:
Duration: 1 to 3 x 50 min: Ocean Literacy

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Principle: #3, #5, #7:

Introduction to Ocean Zones Students will learn to identify and describe different zones of the ocean and the organisms that live in each zone.

Ocean Physics Hands-On Activities

To quantify ocean acidification, the “basic first step” is to measure the pH and ‘aragonite saturation state’ of seawater. The

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saturation state measures the potential for aragonite (a common form of calcium carbonate) to dissolve, and this is an important indicator of how well a coral can calcify and form a skeleton.

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