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2 are only the wavelengths of light that
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Answer Key

Electron Energy and Light 1 Electron Energy and Light How does light reveal the behavior of electrons in an atom? Why? From fireworks to stars, the color of light is useful in finding out what's in matter. The emission of light by hydrogen and other atoms has played a key role in understanding the electronic structure of atoms.

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Electron Energy and Light 3 11. Consider the hydrogen spectrum in Model 2. a. Which color of light has the most energy? b. Which color of light has the least energy? 12. Does a gas discharge tube filled with boron emit the same

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wavelengths of light as a tube filled with hydrogen? Use evidence from Model 2 to support your answer. 13.

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Model 1 - White Light Color	Photon Energy ($\times 10^{-21}$) (J)	Wavelength Range (nm)	Speed (m/s)
Reds	269-318	625-740	3.00×10^8
Oranges	318-337	590-625	3.00×10^8
Yellows	337-352	565-590	3.00×10^8
Greens	352-382	520-565	3.00×10^8
Blues	382-452	440-520	3.00×10^8
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of the following sentences describes the relationship between wavelength and energy of light? (#6c) answer choices .
As wavelength gets shorter the energy
...

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