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2009, Prentice- Hall, Inc. Rate Laws • A rate law shows the relationship between the reaction rate and the concentrations of reactants. • The exponents tell the order of the reaction with respect to each reactant. • Since the rate law is $\text{Rate} = k [\text{NH}_4^+] [\text{NO}_2^-]$ the reaction is. First-order in $[\text{NH}_4^+]$ and. First-order in $[\text{NO}_2^-]$.

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